

# **Evaluation of 2, 2-dichlorovinyl dimethyl phosphate (Sniper) induced hepatotoxicity and oxidative stress in New Zealand white rabbits**

## **ABSTRACT**

**Aim:** The aim of this study was to evaluate chronic hepatotoxicity and oxidative stress induced by 2, 2-dichlorovinyl dimethyl phosphate (Sniper) in New Zealand white rabbits.

**Study design:** This is an experimental study.

**Place and Duration of Study:** Department of Biological Science, Rivers State University, Port Harcourt animal house, Rivers State Teaching Hospital and Nigerian National Petroleum Corporation Hospital Laboratory, between January, 2020 and April 2020.

**Methodology:** Thirty six (36) male New Zealand white rabbits weighing approximately 1.0mg/kg were used for the study. The rabbits were kept in a spacious and well-ventilated cage at room temperature, under natural circadian rhythm and were allowed to acclimatize for fourteen (14) days. They were divided into three (3) groups of four (4) rabbits each with four (4) matched control. For the chronic oral study, 10% of the LD50 (details not included) which is 0.005mg/kg dose of sniper, mixed with 1.0ml of distilled water was administered orally to the rabbits daily for the stipulated period of 0-30, 0-60 and 0-90 days. The matched control rabbits received only feed and water *ad libitum* during the study. Whilst, for the chronic inhalation study, 10% of the LD50 dose of sniper which is equivalent to 0.05mg/m<sup>3</sup> dose of sniper was mixed with 1.0ml of distilled water, sprayed in the closed cages. At day 30, 60 and 90, 4 rabbits were sacrificed each from the chronic oral and inhalation study groups and the matched control group. Blood specimens were collected at each stage, about 5.0mls of blood was collected into lithium heparin specimen container for the investigation of liver function tests. Serum alanine aminotransaminase (ALT), aspartate aminotransaminase (AST), alkaline phosphatase (ALP), total and conjugated bilirubin, (TB and CB) total protein (TP), albumin (ALB), lactose dehydrogenase (LDH), malondealdehyde (MDA) and total antioxidant capacity (TAC) were estimated using the spectrophotometric method. SPSS version 22.0 was used for statistical analysis and p values less than .05 were considered statistically significant.

**Results:** The results showed that there were significant increases in the activity of ALT, AST, ALP, MDA and LDH (at  $p < .05$ ), when values for the controls were compared with those of rats administered with sniper. Significant decreases (at  $p < .05$ ) were also observed in TP, ALB and TAC levels after sniper administration.

**Conclusion:** Chronic inhalation and oral administration of sniper led to hepatotoxicity and oxidative stress, toxic effects were more in rabbits administered through the oral route and increased with period of administration.

**Keywords:** 2, 2-dichlorovinyl dimethyl phosphate (Sniper), hepatotoxicity, New Zealand white rabbits

## **1. INTRODUCTION**

Dichlorvos is a household and public health insecticide with fumigant action, dichlorvos has wide spread use in the form of aerosol or liquid sprays or as impregnated cellulosic, ceramic or resin strips, especially against flies and mosquitoes. For the control of fleas and ticks on livestock and domestic animals (pets), impregnated resin collars are used. A granular form of an impregnated resin strip is used as an antihelmintic in domestic animals. Dichlorvos is

also available as an aerosol and soluble concentrate [1]. Commercial production of dichlorvos began since 1961, and it has been available to household since 1964. It can be produced by the dehydrochlorination of trichlorfon in aqueous alkali at 40 -500C [2]; commercial production is by a reaction of trimethyl phosphate and chloral [3].

Dichlorvos is toxic to both humans and animals and it is used indiscriminately due to its affordability and availability. This has led to its use in high frequency for suicide, homicide and other accidental exposures [3]. The United States Environmental Protection Agency has reviewed the safety data of dichlorvos severally and in 1995 they reached a voluntary agreement which restricted the use of dichlorvos in many uses especially for all aerial application [4]. A permissible exposure limit in the Workplace has been set at a concentration of  $1\text{mg}/\text{m}^3$  over an 8-hour workday by the occupational safety and Health Administration (OSHA). It was also stated that dichlorvos exposure at levels of  $100\text{g}/\text{m}^3$  is instantly dangerous to life and health [5]. The international Agency for research on cancer (IARC) has also stated that dichlorvos is a possible carcinogen to humans [6]. Again, the U.S environmental protection Agency (EPA) has also estimated that a lifetime of drinking water which contains 0.1 microgramme of dichlorvos per liter (mg/L) would cause an extra case of cancer in every million people [6].

Dichlorvos self-poisoning has become a very serious problem in Nigerian society, and it has been estimated that about 200,000 people die every year as a result of pesticide self-poisoning [7]. It has also been assumed that estimate of 873,000 cases of suicide has occurred among the world population in the year 2002 [8]. Several cases of death by pesticides in China, Malaysia, Srilanka and Trinidad-Tobago have been reported, and it is estimated that 330,000 people die by intentional intake of pesticides annually. Exposure to dichlorvos could cause acute or chronic toxicity. Inhalation is usually the most common route of dichlorvos toxicity because of its volatility [9]. The main exposure pathway of dichlorvos in human is the inhalation route of exposure. This is because of its current use patterns. Therefore, a chronic inhalation study could help in the assessment of potential risks of dichlorvos [9].

Serious poisoning with dichlorvos affects the central nervous systems which will lead to incoordination, slurred speech, loss of reflexes, weakness, fatigue, involuntary muscle contraction, twitching, tremors of the tongue or eyelids and paralysis of the body extremities and the respiratory muscles [7]. Involuntary urination and defecation could occur in severe cases, as well as irregular heartbeats, Psychosis unconsciousness, convulsions and coma. Furthermore, respiratory failure or cardiac arrest could lead to death [7]. Chronic exposure to dichlorvos may lead to more symptoms such as impaired memory and concentration, severe depression, disorientation, irritability, insomnia and drowsiness, as well as liver toxicity and oxidative stress. An influenza- like condition has also been reported [10]. The aim of this study was to evaluate chronic hepatotoxicity and oxidative stress induced by 2, 2-dichlorovinyl dimethyl phosphate (Sniper) in New Zealand white rabbits.

## 2. MATERIALS AND METHODS

### 2.1 Experimental Animals

A total of thirty six (36), two-month-old New Zealand white rabbits (*Oryctolagus cuniculus*) that weighed averagely 1.0kg were used for this study. The rabbits were purchased from Department of Biological Science, Rivers State University, Port Harcourt animal house. They were used for oral and inhalation chronic studies. The rabbits were kept in a spacious and well-ventilated cage at room temperature, under natural circadian rhythm and were allowed to acclimatize for fourteen (14) days. They were housed in standard cages and allowed access to feed (Top Feed Finisher Mash, Sapele, Nigeria) and water *ad libitum* from the animal house, department of animal and environmental science, Rivers State University, Port Harcourt. All the animals received humane treatment according to the criteria outlined in

71 the Guide for the Care and Use of Laboratory Animals prepared by the National Institute of  
72 Health.

## 73 **2.2 Procurement and administration of Sniper**

74 1 litre of concentrated solution of sniper (DDVP) insecticide 1000EC (which contains  
75 1000mg of 2-2 dichloro vinyl dimethyl phosphate compound was purchased in Nigeria from  
76 Swiss–Nigeria chemical company which is the sole marketing company for sniper in  
77 Nigeria). For the chronic oral study, 10% of the LD50 dose which is 0.005mg/kg dose of  
78 sniper, mixed with 1.0ml of distilled water was administered orally to the rabbits daily for the  
79 stipulated period of 0-30, 0-60 and 0-90 days. The matched control rabbits received only fed  
80 and water *ad libitum* during the study. Whilst, for the chronic inhalation study, 10% of the  
81 LD50 dose of sniper which is equivalent to 0.05mg/m<sup>3</sup> dose of sniper was mixed with 1.0ml  
82 of distilled water, sprayed in the closed cages. The rabbits were transferred into the closed  
83 cages that have been flirited with sniper to spend 4 hours daily before returning them back to  
84 their normal cages.

## 86 **2.3 Experimental Design**

87 Chart 1: The rabbits were divided into three (3) groups of four (4) rabbits each with four (4)  
88 matched control. A total of 20 cages were used for this experiment as shown below:

Duration	Chronic oral study	Chronic inhalation study	Matched control
0-30 days	4	4	4
0-60 days	4	4	4
0-90 days	4	4	4

## 91 **2.4 Sample Collection, Storage and Analysis**

### 92 **2.4.1 Sample collection**

93 At day 30, 4 rabbits were sacrificed each from the chronic oral study group, chronic  
94 inhalation study group and from the matched control group. Blood specimens were collected  
95 at each stage, about 5.0mls of blood was collected into lithium heparin specimen container  
96 for estimation of liver function tests. The liver was harvested and preserved in 10% formalin  
97 for histological examination.

### 98 **2.4.2 Laboratory investigation of parameters**

99 Liver function tests were carried out at Nigerian National Petroleum Corporation (NNPC)  
100 Clinic, Akpajo, Port Harcourt, while the histological study was carried out at Rivers State  
101 University Teaching Hospital, Port Harcourt, Rivers State, Nigeria.

#### 102 ***2.4.2.1 Determination of Aspartate aminotransferase (AST)***

##### 104 ***Principle***

105 Aspartate aminotransferase (AST) catalyzes the transfer of amino group from aspartate to  
106 ketoglutarate, forming oxaloacetate and glutamate. Oxaloacetate reacts with 2,4-  
107 dinitrophenylhydrazine (DNPH) to form 2,4-Dinitrophenylhydrazone which in an alkaline  
108 medium gives a red-brown colour. The absorbance of the colour is directly proportional to  
109 the concentration of the enzyme.

#### 110 ***2.4.2.2 Determination of Alanine Aminotransferase***

##### 111 ***Principle***

112 The principle involved in the determination of alanine aminotransferases is such that alanine  
113 aminotransferase transfers an amino group from alanine to  $\alpha$ -ketoglutarate producing  
114 glutamate and pyruvate. The reaction that produces pyruvic acid is established by coupling  
115 the alanine aminotransferase catalyzed reaction. Oxoacid generated is quantified by

116 coupling oxo-derivatives formed with 2,4-dinitrophenylhydrazine to form an oxo-acid  
117 hydrazine that is viewed as a reddish color in the presence of an alkaline medium.

#### 118 2.4.2.3 Determination of Alkaline Phosphatase (ALP)

##### 119 Principle

120 Alkaline phosphatase (ALP) catalyzes the hydrolysis of the colourless organic phosphate  
121 ester substrate, p-Nitrophenylphosphate, to the yellow coloured product p-Nitrophenol and  
122 phosphate. The absorbance of the coloured product is directly proportional to the  
123 concentration of the enzyme.

#### 124 2.4.2.4 Determination of Lactate Dehydrogenase (LDH)

125 Lactate dehydrogenase enzyme was determined using photometric method.

##### 126 Principle

127 Lactate dehydrogenase (LDH) catalyzes the conversion of L – Lactate to pyruvate, NAD is  
128 reduced to NADH in the process. The initial rate of the NADH formation is directly  
129 proportional to the catalytic LDH activity. It is determined by photometric measurement of the  
130 increase in absorbance.

#### 131 2.4.2.5 Quantitative Determination of Total Antioxidant Capacity (TAC)

132 Method: Colorimetric. Catalog Number: E-BC-K136

##### 133 Principle

134 A variety of antioxidant macromolecules, antioxidant molecules and enzymes in a system  
135 can eliminate all kinds of reactive oxygen species and prevent oxidant stress induced by  
136 reactive species. The total levels affect the total antioxidant capacity in the system. Many  
137 antioxidants in the body can reduce  $\text{Fe}^{3+}$  to  $\text{Fe}^{2+}$  can form stable complexes with  
138 phenanthroline substance. The antioxidant capacity (TAC) can be calculated by measuring  
139 the absorbance at 520 nm.

#### 140 2.4.2.6 Determination of Malondialdehyde (MDA)

141 The estimation of malondialdehyde was carried out by photometric method.

##### 142 Principle

143 The principle is based on the quantification of a powerful light – absorbing and fluorescing  
144 adduct in a continuous reaction with thiobarbituric acid (TBA).

#### 145 2.4.2.7 Laboratory estimation of albumin

##### 146 Principle

147 Sulphonphthalein dyes as bromocresol purple or bromocresol green yield with albumin in the  
148 presence of detergents in a blue-green complex suitable for the photometric determination.

149 Albumin concentration (g/l) =  $\frac{A_{\text{sample}}}{A_{\text{standard}}} \times \text{concentration of standard.}$   
150

#### 151 2.4.2.8 Laboratory Analysis of Total Proteins

##### 152 Principle

153 Proteins and peptides, similarly to biuret, react with cupric ions in alkaline solutions to form a  
154 violet complex suitable for the photometric determination.

155 Total protein concentration (g/l) =  $\frac{A_{\text{sample}}}{A_{\text{standard}}} \times \text{concentration of standard}$   
156

#### 157 2.4.2.9 Histological Analysis

158 The liver was harvested for histological analysis, and were fixed in 10% formal saline  
 159 solution. The organs were dissected and representative blocks were taken for histological  
 160 processing each with identifying label in a tissue cassette. The fixed tissue blocks were  
 161 dehydrated through ascending grades of alcohol, de-alcoholised in xylene, infiltrated and  
 162 embedded in molten paraffin wax. Sections were cut at 3µm on a rotary microtome.  
 163 Deparaffinised sections were then stained with the standard haematoxylin and eosin staining  
 164 technique and the slides mounted in DPX. Sections on slide were examined and  
 165 photomicrographs captured with X400 objective lens using the ScopeTek™ device and  
 166 software v1.3.

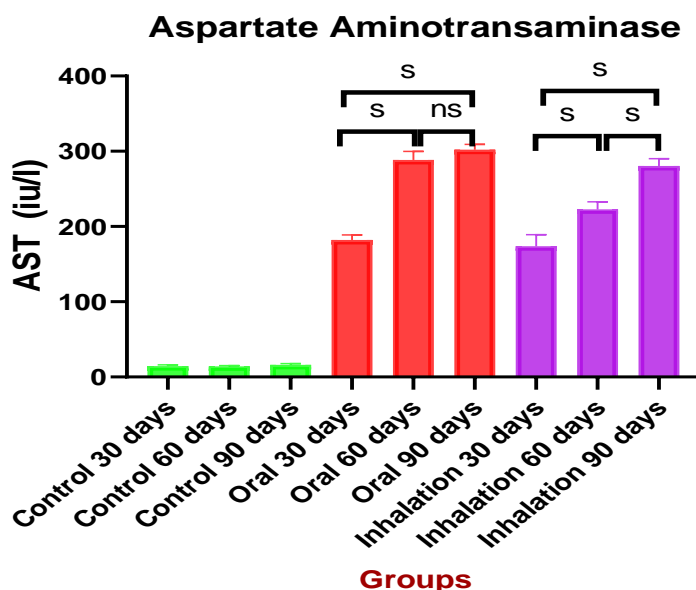
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## 168 2.5 Statistical Analysis

169 SPSS version 22.0 of windows statistical package was used to analyze the data generated.  
 170 The mean ± standard deviation was determined. One way analysis of variance (ANOVA)  
 171 with Tukey's Post Hoc test, bar charts were also done using the same statistical package.  
 172 From the values obtained statistical decision and inferential evaluation were made. A  
 173 probability (p) value of less than .05 was considered statistically significant.

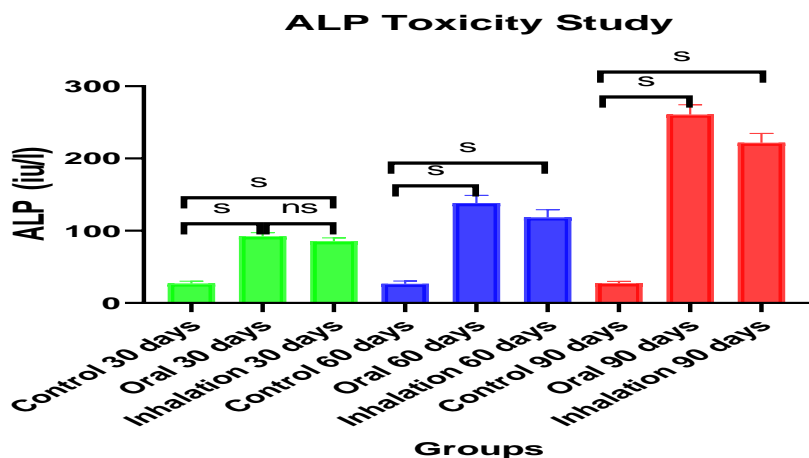
## 174 3. RESULTS AND DISCUSSION

175



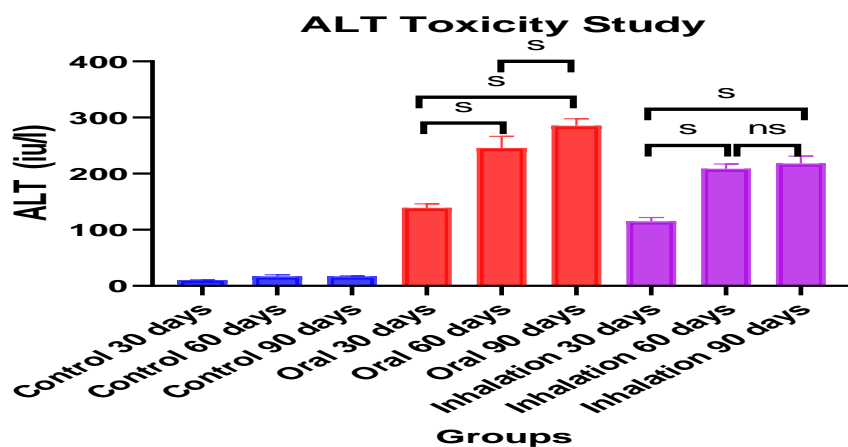
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177 Fig. 1: Serum AST activity comparison of the effect of routes of administration of  
 178 Sniper on liver function parameters



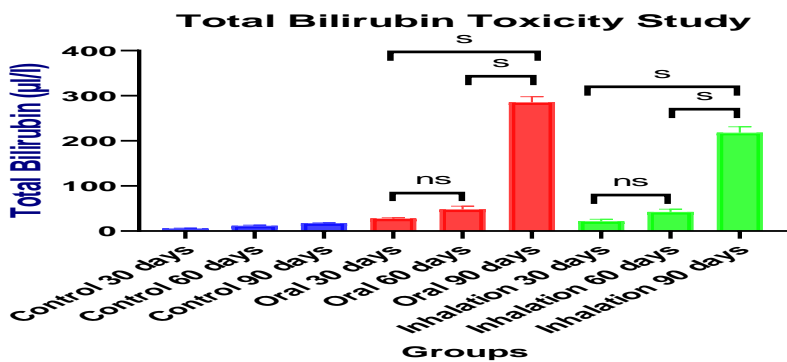
180

181 Fig. 2: Serum ALP activity comparison of the effect of routes of administration of  
 182 Sniper on liver function parameters  
 183



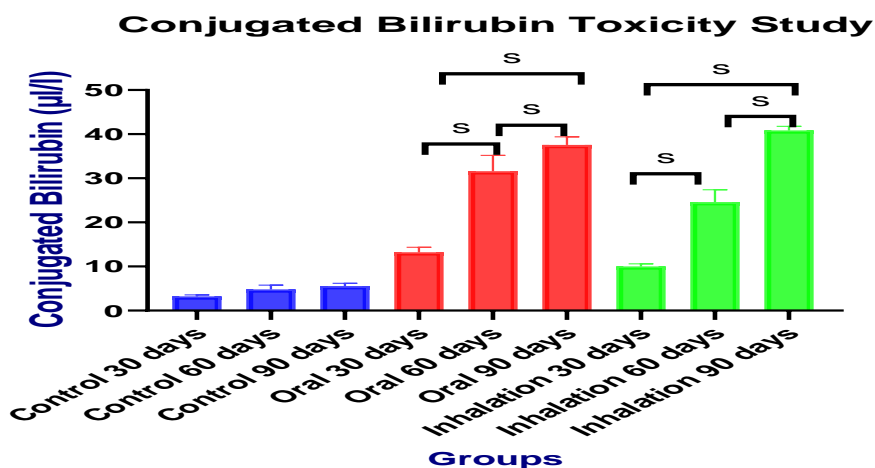
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185 Fig. 3: Serum ALT activity comparison of the effect of routes of administration of  
 186 Sniper on liver function parameters  
 187



188

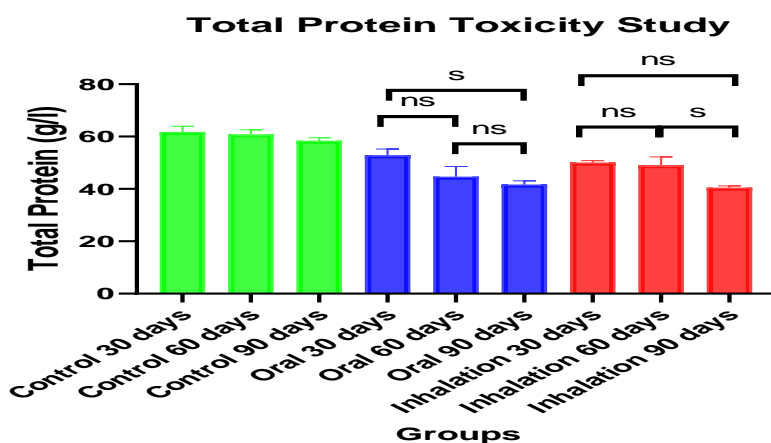
189 Fig. 4: Serum ALT activity comparison of the effect of routes of administration of  
 190 Sniper on liver function parameters



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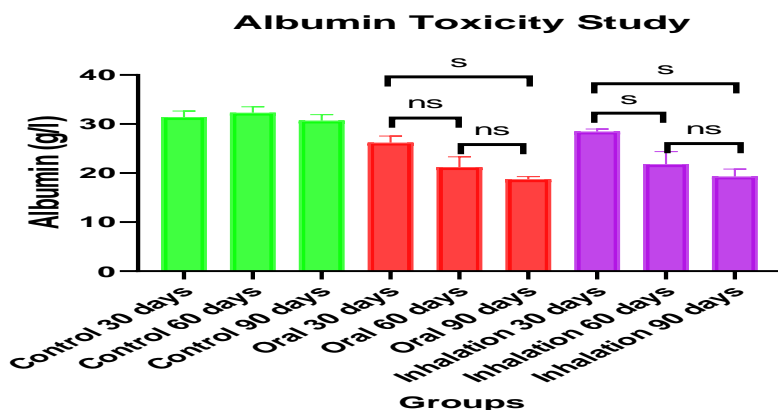
192 Fig. 5: Serum conjugated bilirubin comparison of the effect of routes of administration  
193 of Sniper on liver function parameters

194



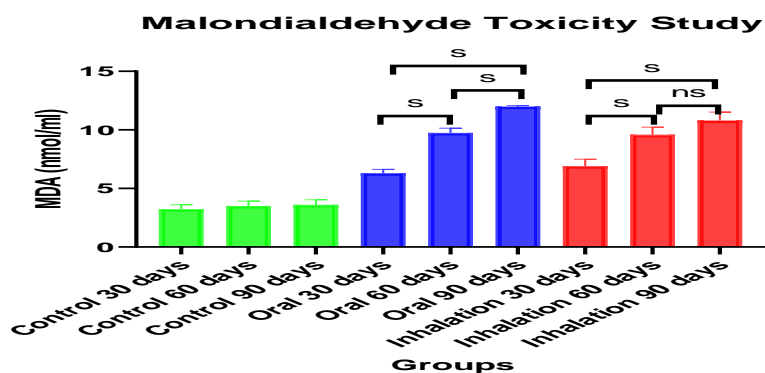
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196 Fig. 6: Serum total protein comparison of the effect of routes of administration of  
197 Sniper on liver function parameters



198

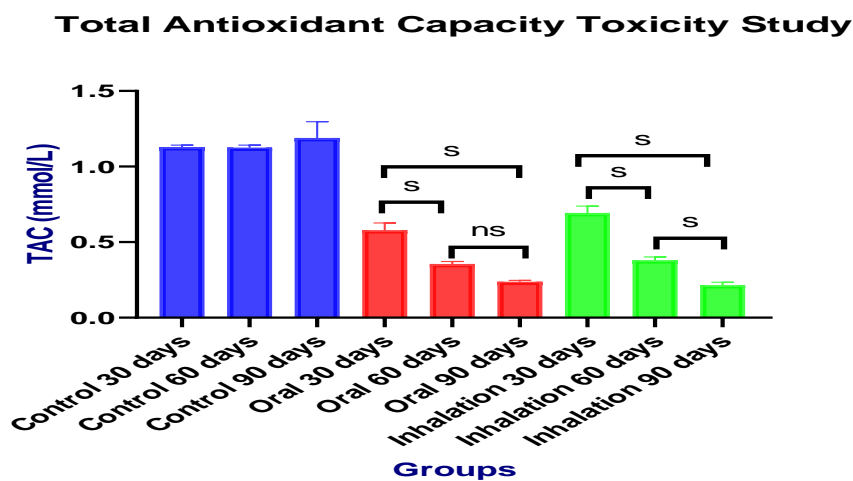
199 **Fig.7: Serum albumin comparison of the effect of routes of administration of Sniper**  
 200 **on liver function parameters**



201

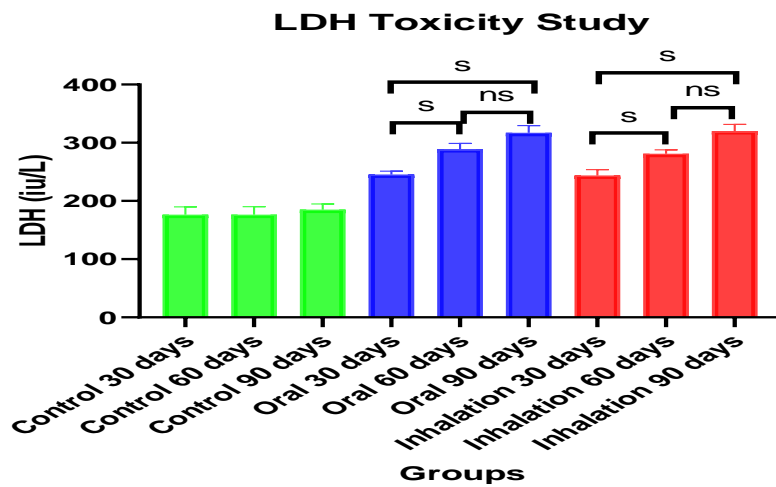
202 **Fig.8: Serum MDA activity comparison of the effect of routes of administration of**  
 203 **Sniper on liver function parameters**

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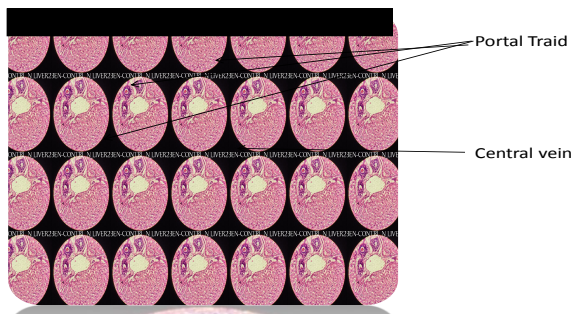
206 **Fig.9: Serum TAC levels comparison of the effect of routes of administration of Sniper**  
 207 **on liver function parameters**



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209  
210

**Fig.10: Serum LDH levels comparison of the effect of routes of administration of Sniper on liver function parameters**

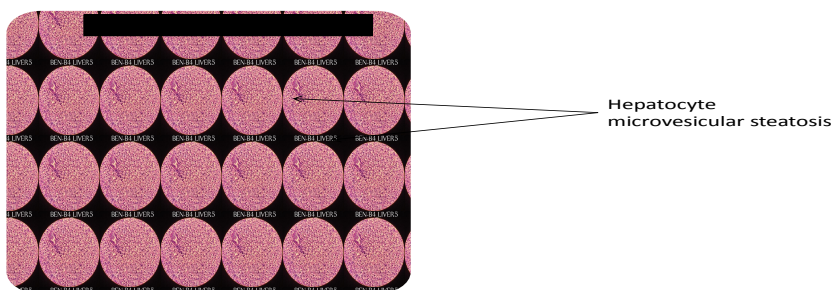
### **Organs of the liver (CONTROL)**



211

**Plate 1: Micrograph of a normal liver (from rabbit in control group- oral)**

### **Oral day 30**



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**Plate 2: Micrograph of a rabbit's liver given oral for 30 days**

## Oral day 60

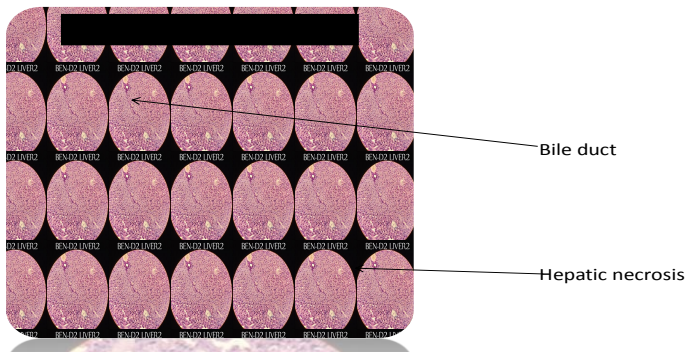


Plate 3: Micrograph of a rabbit's liver given oral for 60 days

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## Oral day 90

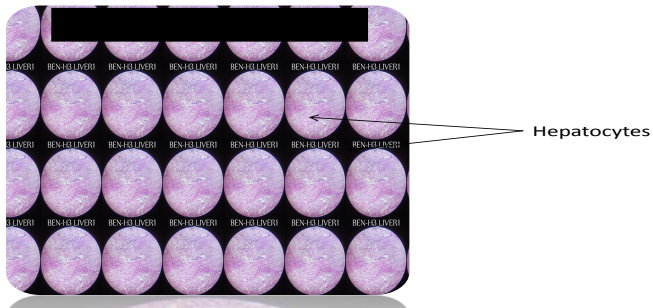


Plate 4: Micrograph of a rabbit's liver given oral for 90 days

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## Inhalation administration (CONTROL)

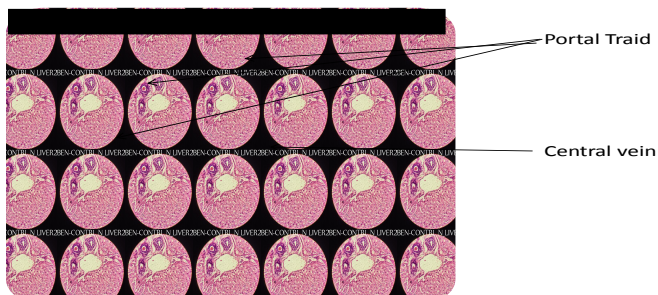


Plate 5: Micrograph of a normal rabbit's liver (control group-inhalation)

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**Inhalation day 30**

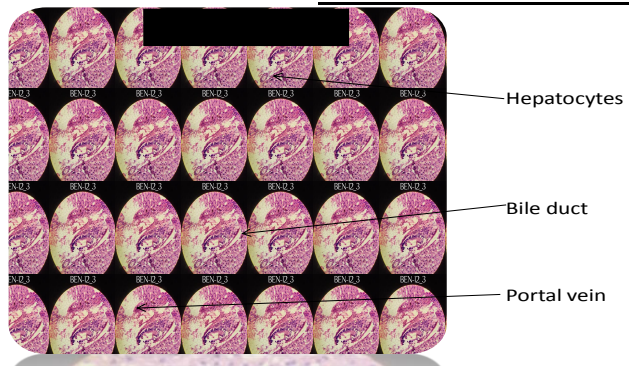


Plate 6: Micrograph of a rabbit's liver given inhalation for 30 days

221  
222

**Inhalation day 60**

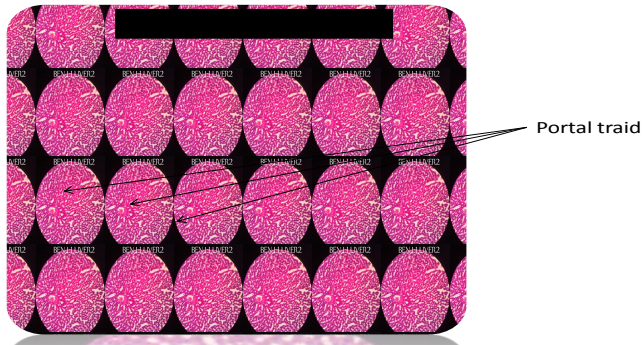


Plate 7: Micrograph of a rabbit's liver given inhalation for 60 days

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## Inhalation day 90

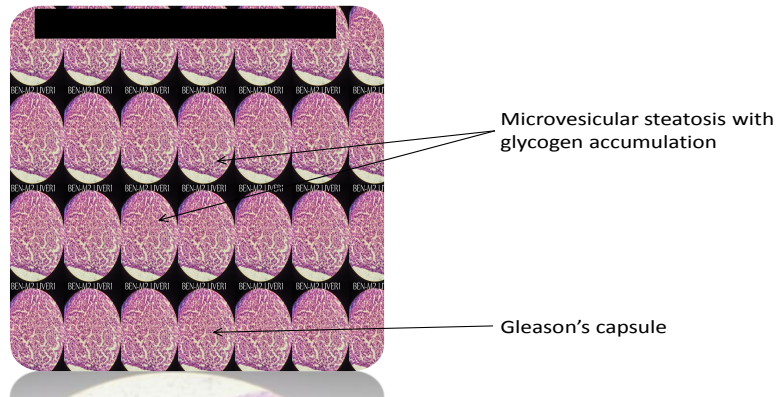


Plate 8: Micrograph of a rabbit's liver given inhalation for 90 days

225  
226

227 This study investigated on the chronic hepatotoxic effects and oxidative stress of inhalation  
228 and oral administration of sniper in New Zealand white rats.  
229 AST, ALT, ALP, total bilirubin, conjugated bilirubin, LDH, TAC, MDA, total protein and  
230 albumin.

231 The liver enzymes are markers of hepatocyte organelles or biliary ductile that leaks out into  
232 the circulation in response to injury. The injury could be caused by reactive metabolites,  
233 resulting from xenobiotic metabolism in the liver. The serum level of these enzymes  
234 therefore reflects physiological state of the liver.

235 Bilirubin, ALT, AST and ALP are present in the blood usually in low levels. When liver cells  
236 are diseased or its integrity disturbed by pathologic conditions, the activity of the enzymes  
237 increases in the blood. The level of these enzymes released into the blood depends on the  
238 severity of the cellular damage; hence the need to monitor these enzymes to ascertain the  
239 state of the liver of the dichlorvos treated rabbits.

240 As shown in Figs. 1 to 10. In the 90 days oral and inhalation dichlorvos treatment of the  
241 rabbits, corresponding increase in the activity of the liver parameters AST (Fig. 1), ALP (Fig.  
242 2), ALT (Fig. 3) and bilirubin (Figs. 4 & 5) was observed as the duration of dichlorvos  
243 treatment on the rabbits increased. The mean value of the liver enzymes was highest in  
244 rabbits that were treated orally for 90 days, followed by the inhalation exposure mean value  
245 of day 90. Generally, comparison of the effect on routes of administration of the dichlorvos  
246 on liver function parameters showed that in the orally treated rabbits, the liver enzymes  
247 activities were slightly higher when compared with the mean values of the inhalation route  
248 exposure.

249 As shown in (Figs. 1 to 10), the day 90 oral and inhalation dichlorvos study produced  
250 significant or remarkable elevation in all the liver function parameters at  $P < .05$ , as the  
251 duration of dichlorvos exposure on the rabbits increased from 30-90 days when compared  
252 with the control.

253 Dichlorvos primarily affects the nervous system of the exposed organisms by inhibiting  
254 acetyl cholinesterase (AChE) and increasing acetylcholine levels in the cholinergic synapse.  
255 Again, dichlorvos induces oxidative stress, causes disturbance of metabolic pathways, and  
256 results in multiple organ dysfunctions such as hypoxia and inadequate tissue perfusion of  
257 the liver and heart [11].

258 In the liver, dichlorvos causes ultrastructural, biochemical, metabolic and mitochondrial  
259 damage which results to changes in the hepatic biomarkers such as serum ALT, AST, ALP  
260 and bilirubin levels as observed in this present study agrees with the findings of [12] who

261 observed increase in the liver function parameters of rats that were exposed to repeated  
 262 doses of organophosphorus pesticides.  
 263 AST and ALT are usually released from the liver when parenchymal cells of the liver become  
 264 necrotic due to viral infection or cell destruction due to exposure to toxic substances. ALT  
 265 enzyme is more specific for the liver than the AST due to their biological location. AST may  
 266 be elevated in other pathological conditions. AST may be elevated also in extra hepatic  
 267 disease.  
 268 The activities of the liver enzymes were markedly raised ( $P<.05$ ) from day 30-90. The  
 269 enzymes were greater than three folds increase when compared with the control. The  
 270 marked increases in the plasma level of the enzymes was an indication of hepatotoxicity,  
 271 which may also indicate degenerative changes and mal-functioning of the liver. The liver cell  
 272 membrane and cytolitic dysfunction by dichlorvos could also be the reason for enzyme  
 273 leakages into the plasma as revealed by the elevated activities. This present study has  
 274 shown that dichlorvos has a harmful effect on the hepatic tissue.  
 275 Furthermore, the marked increase in the liver enzyme activities in the oral and inhalation  
 276 dichlorvos treated rabbits may be attributed to oxidative stress effect of dichlorvos on the  
 277 liver cells. The elevation of AST and ALT along with ALP may reflect some inflammatory  
 278 conditions or injury to the liver (hepatocellular disease). In the present study, more than 3-  
 279 folds of increase in the liver enzyme levels of ALT, AST and ALP in the oral and inhalation  
 280 study at day 90; indicating hepatocellular damage. The result of this study agrees with the  
 281 finding of [13] who observed marked increase in AST, ALT and ALP in a 49 year old woman  
 282 who was chronically exposed to dichlorvos, who was diagnosed of dichlorvos induced  
 283 immune hepatitis on initial hospital admission.  
 284 As shown in (Figs. 4 & 5), there was a significant increase in the mean values of the  
 285 conjugated and total biliubin levels in the oral and inhalation study as the duration of  
 286 dichlorvos exposure prolonged. More than 3-fold increases in the conjugated and total  
 287 bilirubin was observed at day 90 dichlorvos exposure. The hyperbilirubinaemia could  
 288 suggest an over production of bilirubin due to excessive breakdown of red cells. Excessive  
 289 red cells breakdown could be due to antigen antibody reaction caused by several toxins and  
 290 metabolites that were released from the dichlorvos. The elevated levels of conjugated and  
 291 total bilirubin, alongside with the liver enzymes also suggest hepatocellular damage. This  
 292 result corroborates with the findings of Owioye, [12] who also indicated increased levels of  
 293 conjugated and total bilirubin in rats that were exposed to dichlorvos.  
 294  
 295 Serum total protein and albumin were also used in this study to assess the effect of  
 296 dichlorvos on the liver function. The oral and inhalation chronic dichlorvos exposure on the  
 297 rabbits caused an exposure duration dependent decrease in total protein and albumin levels  
 298 at ( $P<.05$ ) from the period of 30-90 days (Figs. 6 & 7). Significant reduction in the levels of  
 299 serum protein and albumin were observed from day 30-90. The both routes of exposure  
 300 produced similar effects. Albumin concentration reflects the balance between synthesis and  
 301 degradation and may be influenced by factors other than the functional state of the liver. It  
 302 can be suggested that the depression of serum total protein and albumin levels which were  
 303 observed in this study may be due to the suppressive action of Metabolites of dichlorvos on  
 304 the synthetic capacity of the liver and or the increase of the renal tubular excretion of protein  
 305 which is lost in the urine. The synthesis of albumin occurs exclusively in the liver and the  
 306 synthetic rate is influenced by thyroid hormones, glucocorticoids, plasma colloidal osmotic  
 307 pressure, hepatic function and toxins.  
 308  
 309 The parameters used in the assessment of the effects of DDVP on the oxidative stress  
 310 markers were total antioxidant capacity and lipid peroxides index. They were used in the  
 311 male rabbits at the various months of exposure. Total antioxidant capacity and lipid  
 312 peroxides index can be used to assess cellular injury which could result in the generation of  
 313 reactive oxygen species. Oxidative stress usually occur due to the production of damaging  
 314 reactive oxygen species (ROS) which is beyond the capacity of the body's natural  
 315 antioxidant defenses, resulting in cellular destruction [12]. Furthermore, oxidative stress is a  
 condition which is characterized by elevated levels of intracellular reactive oxygen species

316 (ROS) which are the progenitors of free radicals. Free radicals are highly reactive, and are  
317 capable of damaging almost all types of biomolecules (proteins, lipids, carbohydrates and  
318 nucleic acids) [14]. Oxidative stress effect on protein results in site-specific amino acid  
319 modifications, fragmentation of the peptide chain, aggregation of the cross-linked reaction  
320 products, altered electrical change and increased susceptibility to proteolysis. Oxidative  
321 attack on iron-sulphur centers in biomolecules by superoxides will lead to the destruction of  
322 enzymatic functions [15].

323 Total antioxidant capacity is the estimate of antioxidant capacity which includes the  
324 antioxidants that are not yet recognized or not easily measured. It represents the overall free  
325 radical scavenging ability of various antioxidants. It is an important measurement for  
326 investigating oxidative stress, which has been implicated in the pathological mechanisms of  
327 many diseases. It measures both the antioxidant capacity of a single compound, and the  
328 antioxidant capacity of all antioxidants in a biological sample. The administration of DDVP on  
329 the male rabbits caused a significant reduction in the level of total antioxidant capacity (TAC)  
330 when the results obtained from the experimental animals were compared with the matched  
331 control values at ( $P<.05$ ) (Figs 8 & 9). Sniper causes damage to the tissues by inducing  
332 oxidative stress, which leads to an increase in the reactive oxygen species in the body,  
333 thereby, depleting the antioxidant enzymes. It has the ability of increasing the production of  
334 reactive oxygen species (ROS) due to its abilities to increase lipid peroxidation, thereby  
335 inhibiting the activity of antioxidant production such as glutathione peroxidase, (GPX),  
336 superoxide dismutase and catalase.

337

#### 338 **4. CONCLUSION**

339 Chronic inhalation and oral administration of sniper led to hepatotoxicity and oxidative stress,  
340 toxic effects were more in rabbits administered through the oral route and increased with  
341 period of administration.

342

#### 343 **ACKNOWLEDGEMENTS**

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346 also wish to thank Dr. U. A. Obisike for his critical & detailed statistical analysis (graphical  
347 presentation) of the results.

348

349 **COMPETING INTERESTS**

350 Authors have declared that no competing interests exist.

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352 **AUTHORS' CONTRIBUTIONS**

353 Authors ESB, EON and FUI designed the study, author ESB performed the statistical  
354 analysis, Author CCO wrote the protocol, and wrote the first draft of the manuscript and  
355 managed the literature searches. All authors read and approved the final manuscript.

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357 **ETHICAL APPROVAL**

358 Principles of laboratory animal care" (NIH publication No. 85-23, revised 1985) were  
359 followed, as well as specific national laws where applicable. All experiments have been  
360 examined and approved by the appropriate ethics committee.

361 **COMPETING INTERESTS DISCLAIMER:**

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363 Authors have declared that no competing interests exist. The products used for this research  
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