Review Article

Revealed State-of-the-Art Review of Phytochemical Content

and Application from Cestrum nocturnum: Comparative

Study for Physicochemical Aspect

ABSTRACT

Cestrum nocturnummostly known as Raat rani, lady of the night as well as night Jessamine which has a great

medicinal impact. The plants that grow in tropical and sub-tropical environments across the world include the night-

blooming jasmine, or Cestrum nocturnum as it is knownbotanically. Covering its fragrant and dazzling white flowers,

Cestrum nocturnum is a well-famed ornamental plant. It's grown as a medicinal plant and used as a hedge. Both

traditional extraction and microwave-assisted extraction can be used to extract the plant. It is possible to determine

qualitatively the phytochemicals, including proteins, amino acids, glycosides, phenolic compounds and tannins. The

point of view of reviews is that extracts can be screened for biological activity and aqueous extract

against E. Coli, B. Subtilis, S. Typhi, S. Aureus, tuberculosis and malaria.

Keywords: Alkaloids, Cestrum nocturnum, Phytochemicals, Steroid, Terpenoid

1. Introduction

Phytochemicals are ingredients derived from plants; chemicals produced by plants during their metabolism

are known as phytochemicals[1]. They generally exhibit biological activity within the plant host and aid

in the growth of the plant or its defence against invaders, diseases or competitors. Since there is currently

insufficient evidence to support their potential health benefits, phytochemicals are typically considered

research compounds rather than necessary nutrients [1], phytochemicals can be divided into three main

groups: flavonoids, lignans, carotenoids and polyphenols, which are found in crude phenolic acids. Based on their similar chemical structures, flavonoids can be further classified as anthocyanins, flavones, flavanones, isoflavones, and flavonols. Proanthocyanins, epicatechins, and catechins are additional classifications for flavanols [2]. More than 25,000 phytochemicals have been found in total. These phytochemicals are typically concentrated in the vibrant sections of plants, such as whole grains, fruits, vegetables, nuts, and legumes. When studying phytochemicals, phyto-chemists first extract and isolate the constituents from the source plant, then define their structure or test them in vitro experiments, in vivo studies involving lab animals, or cell cultures. Difficulties in that area include separating particular compounds, figuring out their frequently intricate structures, and figuring out which particular phytochemical is principally in charge of a given biological activity [2-3]. The phytochemical category involves compounds that are recognized as essential nutrients which are naturally contained in plants and are required for normal physiological functions, so they must be obtained from the diet of humans [3]. Some phytochemicals are recognized phytotoxins that are toxic to humans; for example, aristolochic acid is carcinogenic at low doses. Some phytochemicals are antinutrients that interfere with the absorption of nutrients such as some polyphenols and flavonoids might be pro-oxidants in high ingested amounts [3]. Phytochemicals are a wide variety of nonnutritive chemical compounds found in plant foods which may have health effects. A few examples of well-known phytochemicals are flavonoids, phenolic acids, isoflavones, curcumin, isothiocyanates and carotenoids [4-6]. A species of plant in the potato family Solanaceae is called Cestrum nocturnum, also known as the lady of the night, night-blooming jessamine, night-scented jessamine, night-scented cestrum, or poisonberry. Though it was naturalized in South Asia, its origins are in the West Indies. Although Cestrum nocturnum is thought to be native to America, it has been cultivated for its potent flower scent and has spread throughout the new and old-world tropical regions, particularly the Pacific. It has even become invasive in many of these areas. It has been noted as widespread throughout Mesoamerica (gentry and d'Arcy, 1986)[7]. The species was previously thought to be native to only South America. While Acevedo-Rodriguez and Strong (2012) listed the species as exotic, PIER (2014) and USDA-ARS (2014) listed it as native to Cuba, and Hanelt et al. (2001) merely noted that it was a cultivated species. The species is widely distributed throughout the Pacific region of the Old World's tropics and is known to be invasive on numerous islands. According to reports, it is only grown for ornamental purposes in China and Singapore, and it is a common ornamental cultivation in India [8-9]. *Cestrum* nocturnum is a woody, evergreen shrub that can reach a height of 4 meters (13 feet). Simple, narrow, lanceolate leaves with an entire margin that are smooth and glossy measure 6–20 cm (2.4–7.9 in) long and 2–4.5 cm (0.79–1.77 in) wide. The greenish-white flowers have a thin, tubular corolla. Produced in cymose inflorescences, they are 1-2.5-2.5 cm (0.79-0.98 in) long with five acute lobes, and 10-13 mm (0.39-0.51 in) in diameter when open at night. It releases a strong, sweet scent at night[9]. The fruit is a berry that is either auberge-colored or Marfil white, measuring 10 mm (0.39 in) in length and 5 mm (0.20 in) in diameter. A variety of yellowish flowers are also present. Regarding the toxicity of fruit and foliage, reports vary[10-11].

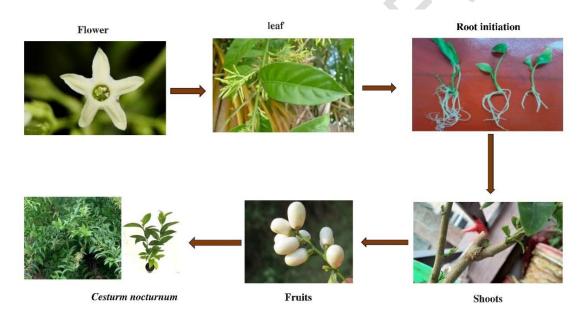


Fig. 1. Life cycle of Cestrum nocturnum

The life cycle of *Cestrum nocturnum* are shown in Fig. 1,[11-13]. Many active ingredients found in the leaves have been used worldwide as a treatment for tropical psoriasis. Practitioners of Ayurveda have also begun to express interest in the plant. Its leaves are applied externally to relieve skin patches, psoriasis, and itching. Malaria can be effectively treated with its oil in many African nations [11-12]. Despite being poisonous, leaves contain several active substances with therapeutic qualities. Additionally, the plant exhibits cardioactive and larvicidal properties [12-13]. Its leaves have long been used to treat burns and swellings. Because of its volatility and ability to ward off mosquitoes, the oil has been used to treat

malaria in numerous African nations. Epilepsy is also treated with it. Pharmacological research has demonstrated that it possesses analgesic, diuretic, antiviral, and abortive qualities and to treat arterial hypertension [12-14]. These contain source of ingredient of essence oil such as Phenyl ethyl alcohol (27%), benzyl alcohol (12%), caryophyllene oxide (3.1%), eicosane (5.6%), eugenol (5.6%), n-tetracosane (4.4%), 1-hexadecanol (2.7%), methoxy eugenol (2.45%), and benzaldehyde (2.32%) are all present in flowers distilled oil[14]. Cytotoxic steroids are present in flower alcohol extract [15-18].

2. Materials and Methods

2.1 Plant Collection and Identification

Fresh plant samples can be collected from Islamic University, Kushtia, Bangladesh. Species identification can be accomplished by comparison with verified herbarium specimens, which can then be verified using diagnostic keys and morphological descriptions found in different floras. The portions that will be useful for the investigationleaves, stems etc will be separated and stored.

2.2 Anatomical Study

After being trimmed to the proper dimensions, the stems and leaves of both plants can be fixed in F.A.A. (formalin, acetic acid-alcohol 1:1:18). Hand sections require a sharp blade for cutting[16]. Safranin and fast green will be used to stain thin transverse sectionswhich will subsequently be passed through alcohol grades to dehydrate them before being mounted in D.P.X. Through the use of a light microscope, observations are made from these sections. Additionally, these sections are photomicrographic. It is necessary to research and identify the plant part(s)' unique identifying characteristics[17].

2.3 Phytochemical Study

After five minutes of washing the leaves in a 5% mercuric chloride solution to get rid of any impurities, they can be shade-dried, baked, and ground into a fine powder [18]. In a rotating evaporator filled with distilled water, ethanol, and methanol, the solvent extracts must be evaporated until they are completely

dry. The resulting dried residues must be stored at -4°C in screw-capped vials [18]. The following techniques could be used to conduct phytochemical studies.

2.3.1 Test for alkaloids

2.3.1.1 Hager's test

Hager's reagent (saturated picric acid solution) must be added to the test solution. The presence of alkaloids will be indicated by the formation of yellow precipitate[19-20].

2.3.1.2. Mayer's reagent and Wagner's Reagent

After warming the plant extract in 2% H₂SO₄ for two minutes, it will filter and a few drops of each reagent will be added individually[21].

- a) Mayer's reagent: There were alkaloids present as a creamy white precipitate [22].
- **b) Wagner's Reagent:** The appearance of a reddish-brown precipitate further indicates that the extract contains alkaloids[22].

2.3.2 Test for Tannins

When the extract is combined with a simple lead acetate solution, the presence of tannins is indicated by the formation of a white precipitate[21].

2.3.3 Test for Cardiac glycosides

A few drops of $FeCl_3$ and concentrated H_2SO_4 are added to the extract solution in glacial acetic acid solution to see the reddish-brown colouration at the intersection of the two layers and the bluish-green colour in the upper layer[23].

2.3.4 Test for saponins

Test for foaming: 20 ml of water is mixed with about 5 ml of filtrate, and the mixture is shaken hard. When standing, a steady fourth will indicate the presence of saponins[24].

2.3.5 Test for Steroids and Terpenoids

0.5 ml of acetic anhydride and 0.5 ml of chloroform will be used to treat 4 mg of extract. After that, a concentrated H₂SO₄ solution must be added[25]. For terpenoids, this will result in a reddish-violet color, and for steroids, a greenish-bluish color[25].

2.3.6 Test for Flavonoids

2.3.6.1. Ferric-chloride test

After adding a few drops of FeCl₃ solution to the extract, it formed a black colour which indicated the presence of flavonoids[26].

2.3.6.2. Lead-acetate solution test

A few drops of a 10% lead acetate solution were added to the extract, causing a yellow precipitate that suggested the presence of flavonoids[27].

c) High-Performance Liquid Chromatography

HPLC (Shimadzu, 2LC-10 ATVP pumps, SPD-10AVP, UV-visible detector, Rheodyne injector with 50 L loop) can be used to analyze methanolic leaf extracts. Software called Shimadzu LC-solution version 6.42 is used to collect and handle the data [28]. The mobile phase will consist of 4% aqueous acetonitrile with 0.1% (v/v) phosphoric acid buffered to pH 3.5 using triethylamine and a Phenomenex C18 column (250mm \times 4.6mm, I.D., 5 μ m). The mobile phase needs to be degassed by sonication for ten minutes after being filtered through a 0.22 μ m membrane filter. The injection volume will be set to 20 μ L, and 260 nm will be used for detection [28].

2.4 Isolation of the essential oil

The air-dried flower parts (200 g) of *Cestrum nocturnum* can be subjected to hydro-distillation for 3.0 h using a Clevenger-type apparatus[28].

2.5 Preparation of organic extracts

The air-dried flower parts (50 g) of *Cestrum nocturnum* can be extracted with n-hexane, chloroform, ethyl acetate and methanol separately at room temperature and the solvents can be evaporated by vacuum rotary evaporator. The extraction process has to yield hexane (7.5 g), chloroform (6.6g), ethyl acetate (5.4 g) and methanol (6.3 g) extracts which are then lyophilized and kept in the dark at 4°C [29-38].

2.6 Antioxidant activity

2.6.1 DPPH Radical Scavenging Activity

The ability of antioxidants derived from natural sources, like fruit and plant extracts, to scavenge free radicals is commonly evaluated using the relatively stable radical DPPH[38]. The capacity of several C. *nocturnum* leaf extracts, at varied concentrations, to scavenge DPPH iterated as a percentage of

inhibition in the Table.2,2-diphenyl-1-picrylhydrazyl, or DPPH for short, is a stable free radical that is frequently used to gauge a substance's antioxidant capacity[38].

% DPPH =
$$\frac{\Delta Absorbance of control - \Delta Absorbance of sample}{\Delta Absorbance of sample} \times 100$$

To evaluate the extracts' capacity to quench DPPH radicals using the technique outlined by Brand-Williams et al. [39]

2.6.2 ABTS Radical Scavenging Activity

2.45 mM potassium persulfate was added to create the ABTS stock solution (7 mM). Before the experiment, the solution was suitably diluted to yield an absorbance of 0.70 at 734 nm. Different concentrations of extract were added to a 100 μL volume [40]. The combinations were added to 900 μL of ABTS solution and incubated for 30 minutes at 37°C. Using ascorbic acid as the standard, the 734 nm absorbance was measured using an Eppendorf Bio-spectrophotometer [40]. The formula utilized to determine the percentage of inhibition was identical to the one employed for DPPH.

3. Characterization

Different techniques were employed to investigate in this review manuscript that is illustrated[16-40].

4. Results and Discussion

Samreen Fatema et. all 2019, studied on phytochemical properties of conventional aqueous extract and microwave-assisted extract of *Cestrum nocturnum*leaves [41-42, 44-49]. It is possible to analyze the phytochemical, physicochemical, and biological characteristics of *Cestrum nocturnum* leaves. The powder sample needs to be treated with various chemicals and registered colour changes. Table 1, presents the findings. Different colours will be produced by the solutions. For instance, the red colour is produced by concentrated hydrochloric acid, while the creamy colour is produced by 1 M hydrochloric acid. It can be explained by variations in the reactivity and colour changes based on the various materials' pH (indicator property). Inorganic compounds were evident in the ash content. It is discovered that Cestrum nocturnum leaves have a total ash content of 13.4%. It was discovered that 50% of the ash was

water soluble and that 22% of the ash was insoluble in 1.0 M hydrochloric acid. The following Table 1 and 2, displays the condensed characteristics of the ash derived from *Cestrum nocturnum* leaves.

Table1: The leave powder shows the following fluorescent test

Sr.	Solutions	Observation	Reference	Sr.	Solutions	Observation	Reference
No.				No.			
1	The powder	Dark green	[41]	7	P +	Cream	[39,41]
	as such (P)				Ammonia		
2	P + n-	Whitish green	[41-42]	8	P + Glacial	Fluorescent	[41, 55,
	butanol				Acetic Acid	green	76]
3	P + conc.	Red	[38, 41]	9	P + 1N HCl	Cream	[41]
	HC1						
4	P + conc.	Dark orange	[41, 43]	10	P + 1N	Yellowish	[41-43]
	HNO_2				NaOH	green	
5	P + conc.	Blackish	[41-42]	11	P + 5% HCl	Creame	[41]
	H_2SO_4	brown					
6	P + Ethanol	Whitish green	[41]	12	P + 5%	Yellowish	[38, 41-
					NaOH		45]

Table2: The leave powder shows the following ash analysis and densities

Sl. No.	Ash	Results	Reference
1	Total ash content	13.5 %	[39, 41]
2	Water soluble content	50.01 %	[40]
3	Acid insoluble content	22.01 %	[40-41]
4	Bulk density content	0.387 g mL ⁻¹	[41]
5	Tab density	0.502 g mL ⁻¹	[41-42]
6	Housner ratio	1.2926	[42]
7	Carrs index	22.6 %	[40-44]

Different techniques were used, but the same solvent was used to extract the leaves of *Cestrum nocturnum*. There was a discernible difference between microwave-assisted extraction (MAE) and conventional extraction (CE). Although the percentages from the two extractions were nearly identical, the MAE is more practical as it provides the same percentage in less than 30 minutes illustrated in table 3.

Table 3: Phytochemical analysis of the leave extract of Cestrum nocturnum

Sr. No.	Reagent	CE	MAE	Reference
	Alkaloids Detec	etion		[42, 46]
i.	Mayer`s test	-ve	-ve	
ii.	Wagner`s test	+ve	+ve	
iii.	Hager`s test	+ve	+ve	

	Carbohydrate Dete	ction		[42-43]
i.	Molish test	+ve	+ve	
ii.	Fehling`s test	+ve	+ve	
iii.	Benedic test	-ve	-ve	
iv.	Barfoad`s test	+ve	+ve	
	Glycosides Detect	cion		[43,45]
i.	Borntrager`s test	-ve	-ve	
ii.	Legal`s test	-ve	-ve	
iii.	Saponins	+ve	+ve	
	Proteins and Amino acid	Detection		[42-43]
i.	Millon`s test	+ve	-ve	4 131
ii.	Nitric acid test	-ve	+ve	
iii.	Biuret test	+ve	+ve	
iv.	Ninhydrine test	-ve	-ve	
	Phenolic compound and tan	nin Detection		[43,47]
i.	Ferric chloride test	-ve	-ve	
ii.	Gelatin test	+ve	+ve	
iii.	Lead acetate test	+ve	+ve	
iv.	Alkaline reagent test	+ve	+ve	

Nishtha et. al. 2023, studied phytochemical analysis and anatomical study of Cestrum nocturnumin Table 3.Different phytochemical constituents analyzed from Cestrum nocturnum are given belowin Table 4. The qualitative analysis revealed the presence of alkaloids, glycosides, saponins and flavonoids in the species[42].

Table 4: Phytochemical studies on the leaf extract of Cestrum nocturnum

Phytochemicals	Aqueous	Ethanol	Methanol	Reference
Alkaloids	+	+	+	[42,49,62]
Glycosides	+	+	+	[43, 48]
Saponins	+	+	+	[43, 69]
Flavonoids	+	+	+	[42, 52]
Terpenoids	+	+	+	[44, 66]
Tannins	-	+	-	[42, 49]
Steroids	+	+	+	[46,48,52]

Amin shaista et.al 2020, studied the Chemical analysis of *Cesturmnocturnum*. To determine the phytochemical study on *Cestrum nocturnum*, the aqueous, methanol and ethanol extracts were used. The phytochemicals like alkaloids, glycosides, saponins, flavonoids, terpenoids, steroids and tannins showed

positive results for all extracts, but tannins showed negative results in the case of aqueous and methanol extracts[42-44].

Table 5: The phytochemical studies on stems, leaves, and flowers of Cestrum nocturnum

Item	V	Various Plant parts		
	Mean value ± SE			
	Stems	Leaves	Flowers	
Total phenolic (mg/gm GAE)	3.01 ± 0.08	3.16 ± 0.07	3.75 ± 0.1	[28, 43, 49,62]
Total flavonoid (mg/gm QE)	1.25 ± 0.03	1.98 ± 0.12	2.19 ± 0.2	[41, 43, 62]
Total tannin (%)	3.21 ± 0.18	3.44 ± 0.25	3.04 ± 0.13	[43, 48, 51]
Total saponin (%)	3.08 ± 0.13	3.03 ± 0.16	2.03 ± 0.03	[5, 43-45]
Total alkaloid (%)	1.32 ± 0.05	1.76 ± 0.07	1.27 ± 0.09	[3, 42-45]

The total antioxidant capacity was highest in leaf extracts, indicating strong antioxidant activity in these extracts, possibly due to the presence of phenolic compounds [45]. Phytochemicals, including alkaloids and flavonoids, found in many plants can exhibit antimicrobial properties expressed in Table 5. These compounds may help protect the plant from pathogens and could have implications for medicinal use in humans [44].

Table 6: The phytoconstituents of various parts of the *Cestrum nocturnum* plant

Sl. No.	Phytoconstituents	Plant Part	Reference
1	Carbohydrates	Flower and	[29, 43]
		Stem	
2	Glycosides: Pregnane glycosides, Cholestane glycosides, a	Leaves	[30, 49]
	Pregnane-Carboxylic acid ç-Lactone glycoside, Nocturnoside		
	A and Nocturnoside B, Phenol glucosides		
3	Triterpenes and sterols: Quassinoids	Stems	[17, 43-52]
4	Coumarins	Aerial parts	[43-53]
5	Alkaloids	Leaves and	[36-42]
		Stems	
6	Flavonoids	Stems	[43-45]
7	Tannins	Stems	[22, 42-43,
			48-52, 67]
8	Volatiles	Flower	[43-45]
9	Saponins: Pseudo-furostanol saponin, Spirostanol saponin,	Stems	[11, 41-46]
	Furostanol saponin,		

The phytoconstituents of various parts of the *Cestrum nocturnum* plant are described in Table 6. The significance of the alkaloids found in leaves, steams and volatiles are presented on flowers.

 Table 7: Various volatile oils from the Cesturmnocturnum plant

Sl.	Compound	Percentage	Odor description	Reference
No.				
1	Acetaldehyde	0.2%	Pungent, penetrating	[31-47]
2	Acetaldehyde diethyl	0.1%	Fresh, fruity green	[30-47]
	acetal			
3	Acetic acid	0.4%	Stinging, sour	[36-43]
4	Acetophenone	0.3%	Pungent, sweet(acacia)	[7, 47]
5	2-acetyl furan	0.4%	Balsamic-sweet	[47]
6	Amyl alcohol	0.1%	Fusel-like	[8, 47]
7	Amyl benzoate	0.2%	Balsamic-sweet	[47]
8	Benzaldehyde	2.5%	Biter almonds	[31, 36]
9	Benzoic acid	0.6%	Odorless, faint-balsamic	[47]
10	Benzophenone	0.4%	Powdery rose, geranium	[6, 47]
11	Benzyl acetate	1.8%	Sweet, floral, fresh	[61]
12	Benzyl alcohol	2.4%	Pleasant, fruity-floral	[36-39]
13	Benzyl butyrate	0.2%	Heavy, fruity-floral	[47]
14	Benzyl phenylacetate	0.4%	Sweet, honey-floral	[47]
15	Borneol	1.3%	Camphor, woody-peppery	[47]
16	Benzyl acetate	1.8%	Ethereal-fruity	[9, 47]
17	Camphor	0.8%	Penetrating, warm-minty	[31]
18	B-caryophyllene	0.3%	Woody-spicy, dry	[11, 47]
19	1,8-cineole	1.4%	Fresh, camphoraceous	[47]
20	Citronellal	0.4%	Lemon, citronella, rose	[31, 36, 47]
21	Citronellol	0.7%	Fresh, rose	[47]
22	Citronellyl acetate	0.6%	Fresh, rosy, fruity	[47]
23	Citronellyl propionate	1.1%	Fresh, fruity, sweet rosy	[39, 62]
24	3-decen-3-one	0.6%	Fruity-floral, jasmine	[36]
25	Dihydrojasmone*	0.9%	Floral, fresh, fruity	[47, 70]
26	Dihydrojasmone lactone*	0.4	Heavy floral, fatty waxy	[47]
27	Hexanol	0.5%	Fatty-fruity	[31]
	Dodecanal	0.3%	Waxy-herbaceous, floral	[12, 47]
29	Ethyl alcohol	0.9%	mild sweet-ethereal	[47]
30	Eugenol	1.3%	Warm spicy, clove	[14, 47]
31	Farnesene*	0.3%	Mild, sweet, warm	[63]
32	Farnesol*	0.2%	Flowery, mild, sweet	[47]
33	Geraniol	0.3%	Green-floral, rose	[31]
34	Geranyl aceate	0.4%	Sweet, fruity-floral, rosy	[47, 74]
35	Heptanal	0.2%	Fatty, harsh, pungent	[64]
36	Heptanal diethylacetal	t	Fresh, herbaceous	[31, 40]
37	Heptanoic acid	0.5%	Fatty-rancid, sweet sour	[55]
38	Heptyl acetate	0.1%	Pleasant, sweet	[13, 47]

39	2-(heptyl)-	t	Fruity-floral	[47]
	tertrahydrofuran			
40	Hexanal	0.6%	Fruity, fatty-green	[47]
41	Linalool	3.1%	Refreshing, floral-woody	[64, 71, 70, 74]
42	Phenylacetaldehyde	0.4%	Harsh-green, hyacinth	[16, 47]
43	Cis-jasmone	2.1%	Fruity fresh, jasmine,	[47]
			warm spicy	
44	Phenol	1.6%	Harsh, pungent	[57]
45	Methyl jasmonate	1.5%	Sweet, floral, jasmine	[15, 65]
46	Neryl acetate	0.4%	Fruity, floral, rosy	[47]
47	Nerol	0.7%	Fresh, sweet, rose	[47, 65]
48	α- phellandrene	9.1%	Fruity	[31]
49	β-phellandrene	12.2%	Floral,sweet	[66]
50	(E)-β-ocimene	9.1%	Refreshing	[17, 47]
51	Octyl acetate	t	Fruity	[19-20, 36]
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As indicated in Table7, the volatile constituents of *C.nocturnum* were inquisition using GC and GUMS combined with GC-sniffing technique. The concentration which is higher than 1% is the main constituents (linalool (3.1%), benzaldehyde (2.5%), benzyl alcohol (2.4%), phenylacetaldehyde (2.4%), cis-jasmone (2.1%), benzyl acetate (1.8%), phenol (1.6%), methyl jasmonate (1.5%), 1,8-cineole (1.4%), borneol (1.3%), eugenol (1.3%), linalyl acetate (1.2%) and citronellyl propionate (1.1%). Here we found many other compounds that are accountable for the beautiful, enchanting odor. An aspect seems that the most abundant compounds contain constituents ranging between 0.1% to 3.0% with linalool at 3.1%. In addition, fatty acids and their esters are responsible for this attractive odour. With the help of the GC-sniffing technique, it was determined that a concentration of more than 1% is responsible for this aromatic characteristic. The jasmine-like aroma also occurred for some of these volatile compounds. This result can be significantly compared with the odour characterization of the single compounds which identified that some showed fresh floral impressions. Overall, there are significant varieties within this study and the anterior inquisition [51-53]. The extraction methods of oils, developmental stages, climate, geographical conditions, and the existence of different varieties of *C.nocturnum* can be responsible for these disparities [43, 74].

Table 8: Antioxidant activity of *Cesturmnocturnum leaf extract*

IC50 values are a measure of the potency of a compound in inhibiting a specific biological activity. In this study, IC50 values were determined for the inhibitory activity of *Cestrum nocturnum* leaf extracts against DPPH and ABTS.

The activity	Plant extract/Standard	$IC_{50} (\mu g/mL)$	Reference
	n-Hexane	185 ± 0.71	[49, 52]
	DCM	NS	[52-53]
DPPH	Ethyl acetate	NS	[53, 68]
	Methanol	39.11 ± 0.53	[52, 65]
	Aqueous	NS	[52, 67]
	Ascorbic acid	15.12 ± 0.7	[46, 52, 72]
	n-Hexane	79.13 ± 0.52	[51, 66]
	DCM	NS	[52, 63-65]
ABTS	Ethyl acetate	50.42 ± 0.76	[57]
	Methanol	20.94 ± 0.85	[54, 68]
	Aqueous	56.73 ± 0.66	[52, 70, 73]
	Ascorbic acid	22.77 ± 0.43	[69]

Table 8, the antioxidant activity of extracts from *Cestrum nocturnum* leaves was assessed using the DPPH assay, with the methanol extract showing the highest potency in neutralizing the DPPH radical, while ascorbic acid served as the reference standard. In the current study, *C. nocturnum* leaves were gradually extracted using n-Hexane, DCM, EtO-Ac, MeOH, and water. The phytochemical screening results show that bioactive compounds like flavonoids and polyphenols are significantly concentrated in the MeOH extract (Table 2), which are recognized for their capacity to squelch free radicals. Reductases function as antioxidants by contributing hydro-free radicals to gen atoms, and their concentration is correlated with the reducing power of the antioxidant properties of extracts [52-55]. The MeOH extract showed noteworthy (Table 5) total antioxidant activity. These outcomes are consistent with those from earlier research [14, 44, 56, 57]. Additionally, a newly released study showed that *C. nocturnum* leaves are abundant in phytochemical components [56].The antioxidant capacity of the MeOH extract may be attributed to poly-phenolic components, as phenolic compounds are thought to account for most of the antioxidant capabilities of plant extracts [58-59]. Certain chronic diseases can be stopped or their course can be delayed by compounds that can quench free radicals and reduce oxidative stress [60-63]. Strong DPPH and ABTS radical quenching ability was demonstrated by the methanol extract of *C.*

*nocturnum*leaves (Table 5), demonstrating the plant's antioxidant activity. These findings concur with previously released reports [64-65].

5. Significance Application Area

Table 9: The specific area illustrated below

Sl. No.	Medicinal uses	Reference
1	It is used in epilepsy treatment.	[66]
2	The leaves had strong bactericidal and analgesic properties.	[36], [67]
3	One uses the volatile oil to ward off mosquitoes.	
4	It is used to treat malaria in African nations.	[31]
5	The plant has an anaesthetic effect, an inhibitory effect on the central nervous system, and a cardiac athymic effect.	[68]
6	It contains many antioxidants including beta-carotene, lutein, zeaxanthin, lycopene, alpha-tocopherol, and phytosterols. These fight against free radicals and cell damage are protected which are protected by oxidation.	[69]
7	It can improve heart health.	[69]
8	The presence of antioxidantshelps to lower cholesterol levels.	[69]
9	It may also help prevent cancer.	[69]
10	It is also known as boost immunity.	[69]
11	It is used for the treatment of allergies, asthma, bronchitis, and sinus problems.	[69]
12	This also promotes sleep because of its aromatic characteristics.	[69]
13	It helps to feel more relaxed and calmer.	[69]
14	It contains volatile oils that are worked as natural insect repellents.	[69]
15	It is also effective against dengue fever, yellow fever, West Nile virus, and encephalitis.	[69]
16	The plant has been used for centuries to treat fever, colds, coughs, flu, sore throat, and respiratory infections.	[69]
17	This flower fragrance and essential oil also help to reduce stress and anxiety.	[69]

18	It is an excellent source of vitamins(A, B, C, D, E, K), iron, magnesium, calcium, potassium, sodium, zinc, copper, manganese, selenium and fibre.	[69]
19	Ethanolic extract from <i>C.nocturnum</i> leaves has a wound-healing effect and is prepared as an ointment.	[69]
20	It has a high rate of wound contraction.	[69]
21	n-butanol and polysaccharides extracts of <i>C.nocturnum</i> can restrict tumor growth.	[70]
22	The dry leaf powder of the plant carried analgesic activity and psychoactive activity.	[70]
23	Green synthesis of silver nanoparticles has strong antioxidant activity along with antibacterial activity.	[70]
24	It has very effective results in diabetes treatment.	[70]
25	It also induced damage to the cancer cell DNA and restricted toxicity.	[70]
26	The volatile oil of this plant and flowers helps to increase serotonin levels in the brain.	[70]
27	Helpful in treating liver disorders, and skin eruptions.	[70]
Sl. No.	Traditional uses	Reference
1	C.nocturnum flowers are used by Hindu religious people for offering their God Shiva and Ganesh in Kathmandu.	[71]
2	Nepalese shamans make ceremonial incense out of leaves and fresh flowers. To strengthen their spiritual healing energies, they consume fresh flowers.	[71]
3	In the West Indies, as a stupefying charm medicine, these plants were used.	[73]
4	The Yucatec Maya used the leaves and flowers in hot baths to treat night sweets.	[73]
5	The Kalinchok region, to the north of Kathmandu, occasionally uses the plant to make liquor.	[73]
6	In traditional medicine, the leaves of these plants are used for their pharmacological significance.	[72]
7	This flower could be used as gajra, or flower garlands in wedding rituals at Tamil Naru and many other states of India.	
8	Used as worship for many religious purposes.	
9	The identification and isolation of the propyl gallate compound from the plant extract using traditional methods suggest its potential application in various	[46]
	industries, such as food preservation	

The utmostapplications are described in Table 9 which expresses the impact of human life as well as other fields of specialization.

6. Conclusions

The fragrant night-blooming jasmine plant has a wide range of therapeutic uses, including wound healing, hepatoprotective, antipyretic, antioxidant, antibacterial, antifungal, anticancer, hypoglycemic, antimalarial, and antiepileptic effects. Because of its stunning and fragrant white flowers, night-blooming jasmine is also used as an ornamental plant. Chinese traditional medicine applies the leaves of C. nocturnum externally to treat swellings and burns. To confirm the effects listed above in humans, more investigation and clinical testing are required. Methods such as screening and phytochemical profiling of these plants aid in the identification of elite species. Correct identification of herbal drugs in commerce aids in the prevention of drug piracy and, as a result, makes genuine botanicals available to consumers and drug manufacturers for their proper use as medicine. The precise amount required to make a more efficient source of medicine for its use in the best way is determined by this process. For phytochemicals such as alkaloids, sugars, glycosides, saponins, proteins, amino acids, phenolic compounds, or tannin content, the extract yields the same result. Additional research can be conducted to evaluate the pharmacological activities and potential applications of the isolated compound, propyl gallate, in various fields such as medicine, agriculture, and cosmetics. Comparative studies can be carried out to assess the phytochemical composition and biological activities of Cestrum nocturnum from different geographical locations to determine any variations in its chemical profile. Additionally, the dosage, preparation methods, and potential side effects of any herbal remedy should be carefully considered. More research is needed to fully understand the pharmacological properties of Cestrum nocturnum and its potential applications in modern medicine.

Data Availability

The research data is available on request.

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